

PHOTOLYSIS

*The driver for
photo-oxidation*

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NCAR

Atmospheric Oxygen

Thermodynamic Equilibrium

Normal O₂ molecules

Ozone, O₃

Ground state atoms, O(³P)

Excited atoms, O*(¹D)

ΔH_f
kcal mol⁻¹

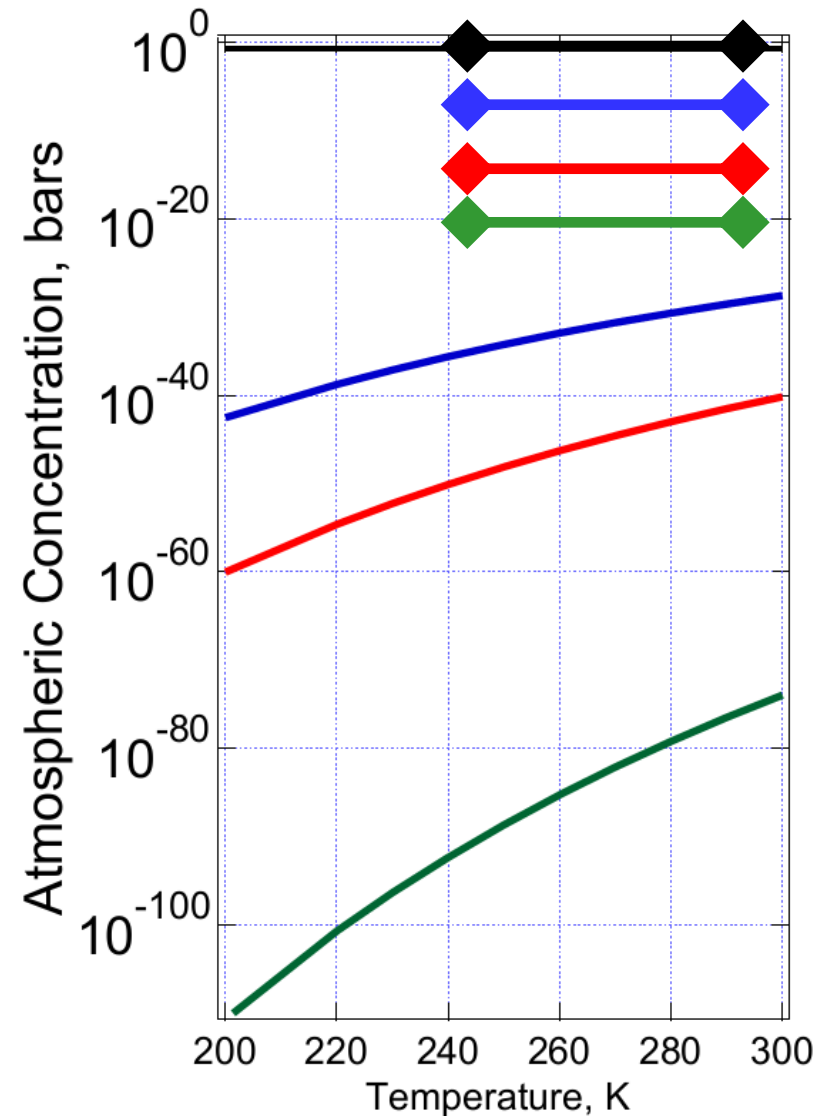
0

34.1

59.6

104.9

observations

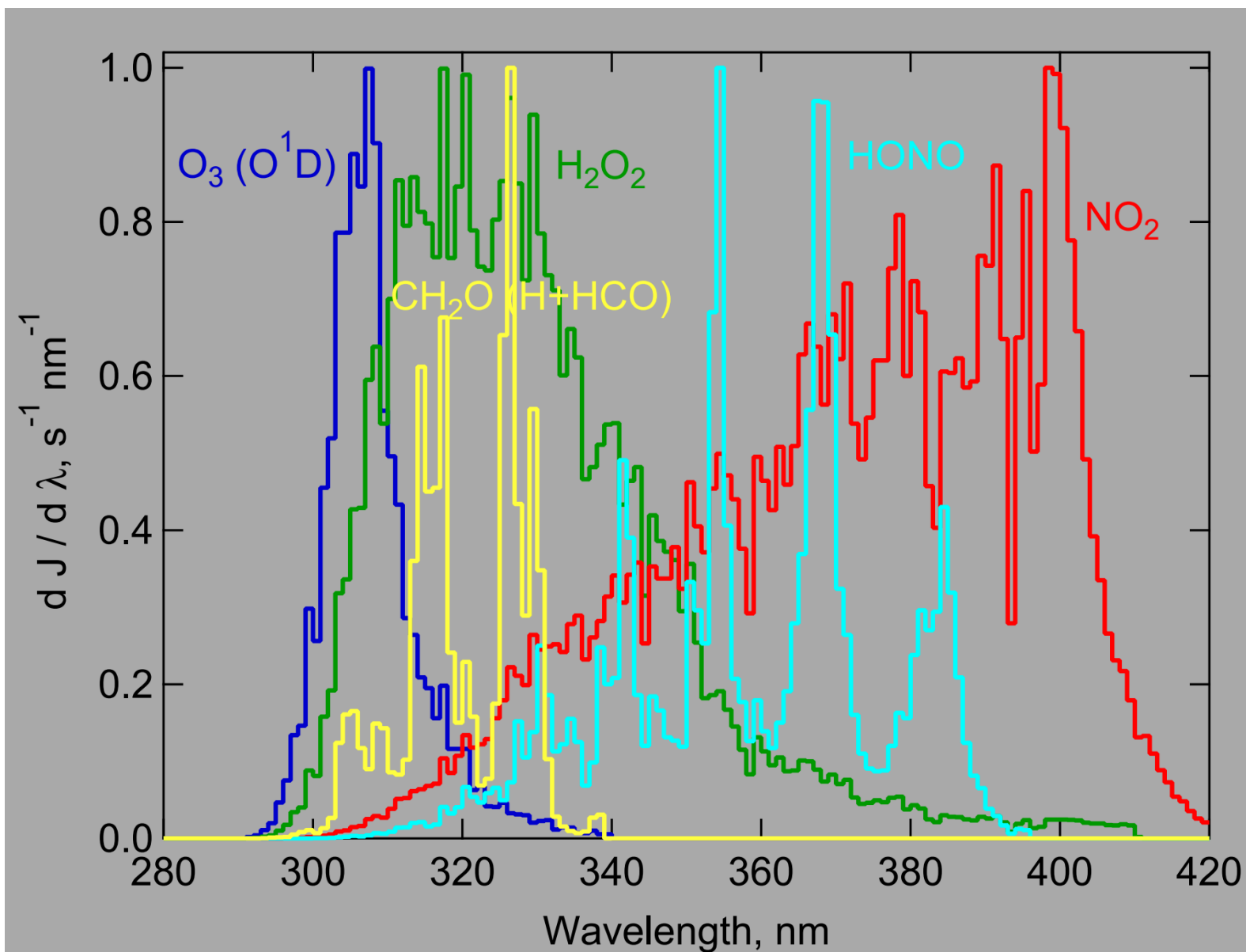


Some Important Photolysis Reactions

$\text{O}_2 + h\nu (\lambda < 240 \text{ nm}) \rightarrow \text{O} + \text{O}$	source of O_3 in stratosphere
$\text{O}_3 + h\nu (\lambda < 340 \text{ nm}) \rightarrow \text{O}_2 + \text{O}(^1\text{D})$	source of OH in troposphere
$\text{NO}_2 + h\nu (\lambda < 420 \text{ nm}) \rightarrow \text{NO} + \text{O}(^3\text{P})$	source of O_3 in troposphere
$\text{CH}_2\text{O} + h\nu (\lambda < 330 \text{ nm}) \rightarrow \text{H} + \text{HCO}$	source of HOx, everywhere
$\text{H}_2\text{O}_2 + h\nu (\lambda < 360 \text{ nm}) \rightarrow \text{OH} + \text{OH}$	source of OH in remote atm.
$\text{HONO} + h\nu (\lambda < 400 \text{ nm}) \rightarrow \text{OH} + \text{NO}$	source of radicals in urban atm.

UV-B and UV-A Wavelength: Range and Resolution for Tropospheric Chemistry

sea level, overhead sun, tuv5.2



Quantifying Photolysis Processes

Photolysis reaction: $AB + h\nu \rightarrow A + B$

Photolysis rates: $\left. \frac{d[AB]}{dt} \right|_{h\nu} = -J[AB]$

$$\left. \frac{d[A]}{dt} \right|_{h\nu} = \left. \frac{d[B]}{dt} \right|_{h\nu} = +J[AB]$$

Photolysis frequency (s^{-1}) $J = \int_{\lambda} F(\lambda) \sigma(\lambda) \phi(\lambda) d\lambda$

(other names: photo-dissociation rate coefficient, J-value)

CALCULATION OF PHOTOLYSIS COEFFICIENTS

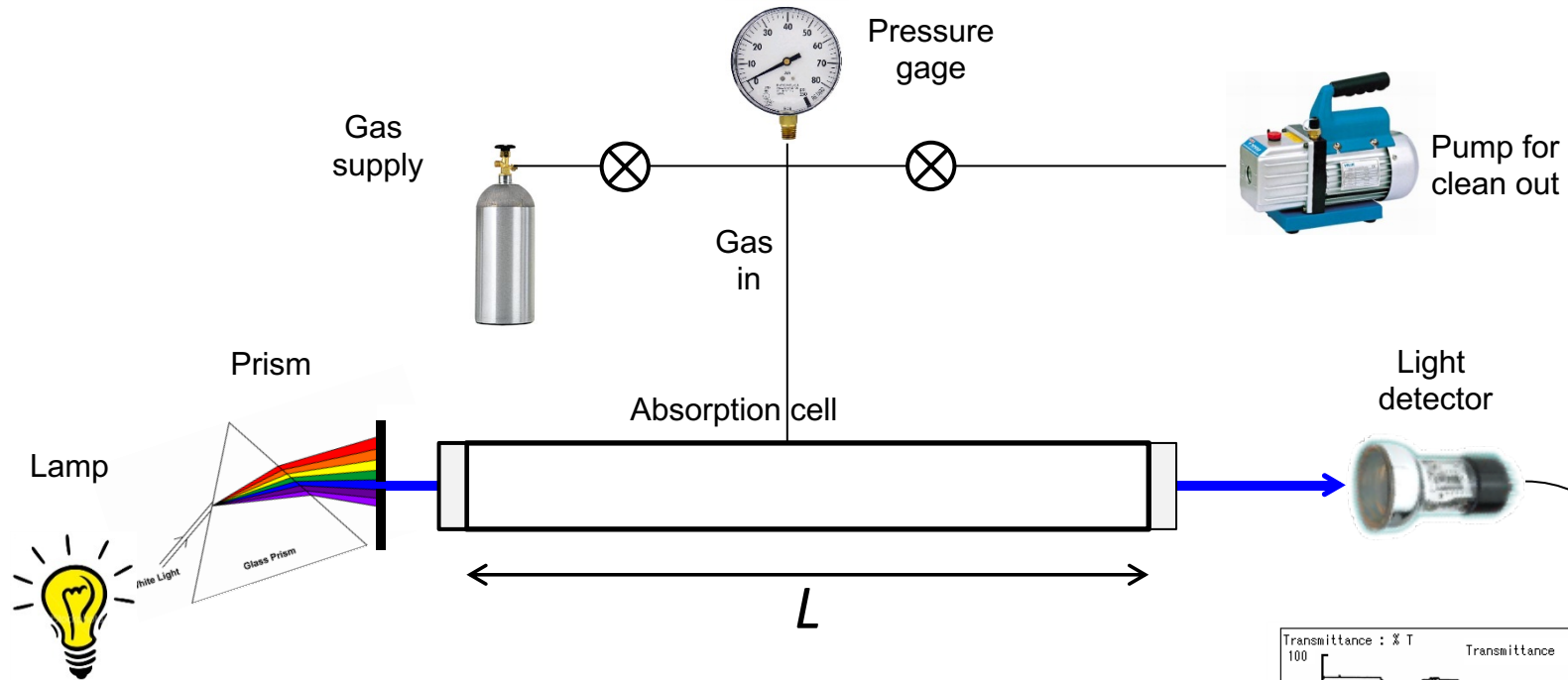
$$J (\text{s}^{-1}) = \int_{\lambda} F(\lambda) \sigma(\lambda) \phi(\lambda) d\lambda$$

$F(\lambda)$ = spectral actinic flux, quanta $\text{cm}^{-2} \text{s}^{-1} \text{nm}^{-1}$
 \propto probability of photon near molecule.

$\sigma(\lambda)$ = absorption cross section, $\text{cm}^2 \text{molec}^{-1}$
 \propto probability that photon is absorbed.

$\phi(\lambda)$ = photodissociation quantum yield, molec quanta $^{-1}$
 \propto probability that absorbed photon causes dissociation.

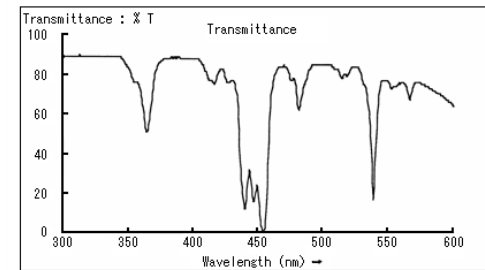
Measurement of Absorption Cross Section $\sigma(\lambda)$



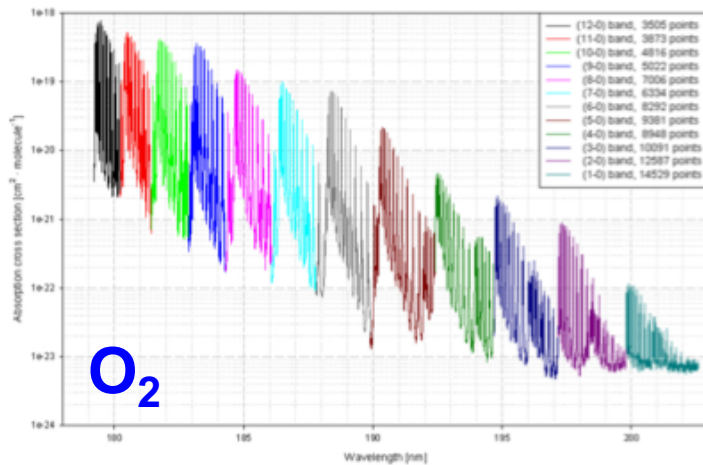
$$\text{Transmittance} = I / I_0 = \exp(-\sigma n L)$$

$$\sigma = -1/(nL) \ln(I / I_0)$$

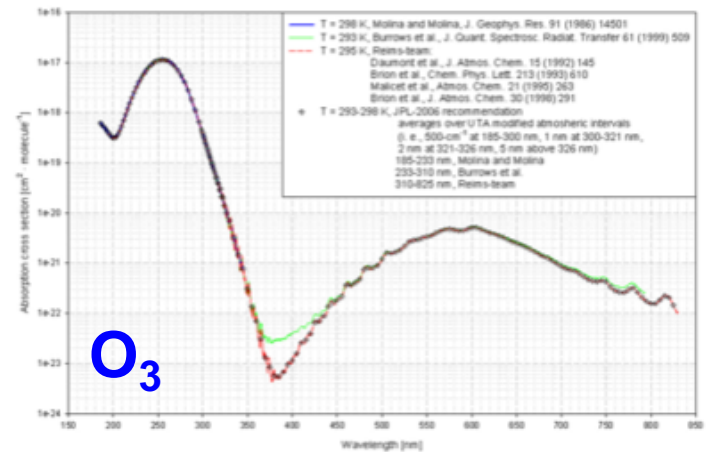
Easy: measure pressure ($n = P/RT$), and relative change in light: I / I_0



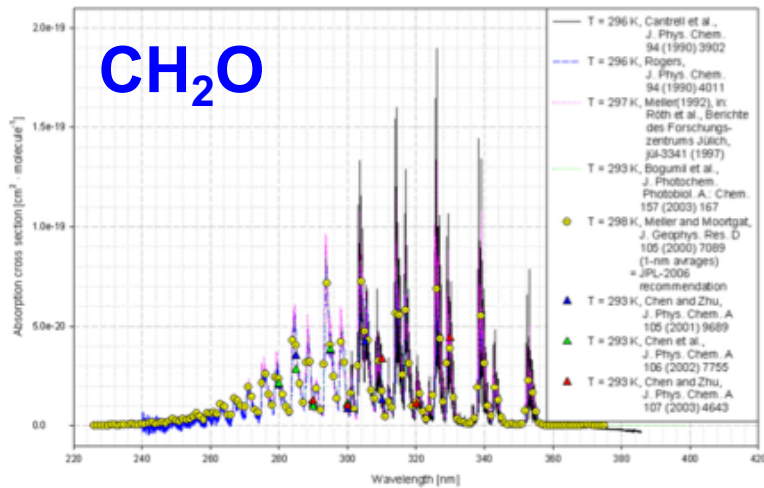
Absorption cross sections $\sigma(\lambda, T)$



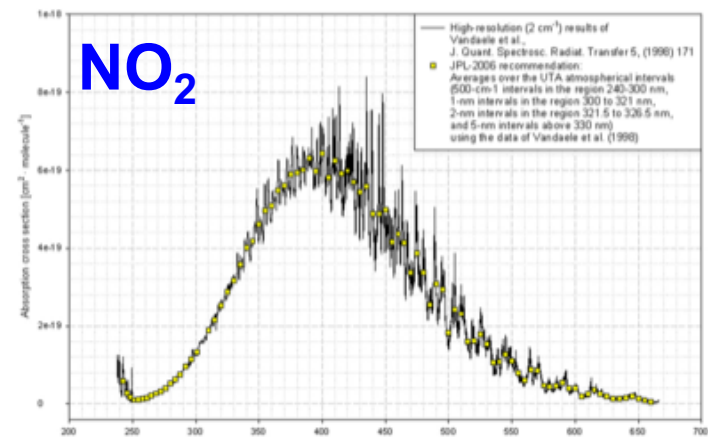
Absorption cross sections in the Schumann-Runge region of oxygen O_2 at 300 K, Yoshino et al., Planet. Space Sci. 40 (1992) 185



Absorption cross sections of ozone O_3 at room temperature
Evaluation for JPL-2006 recommendation

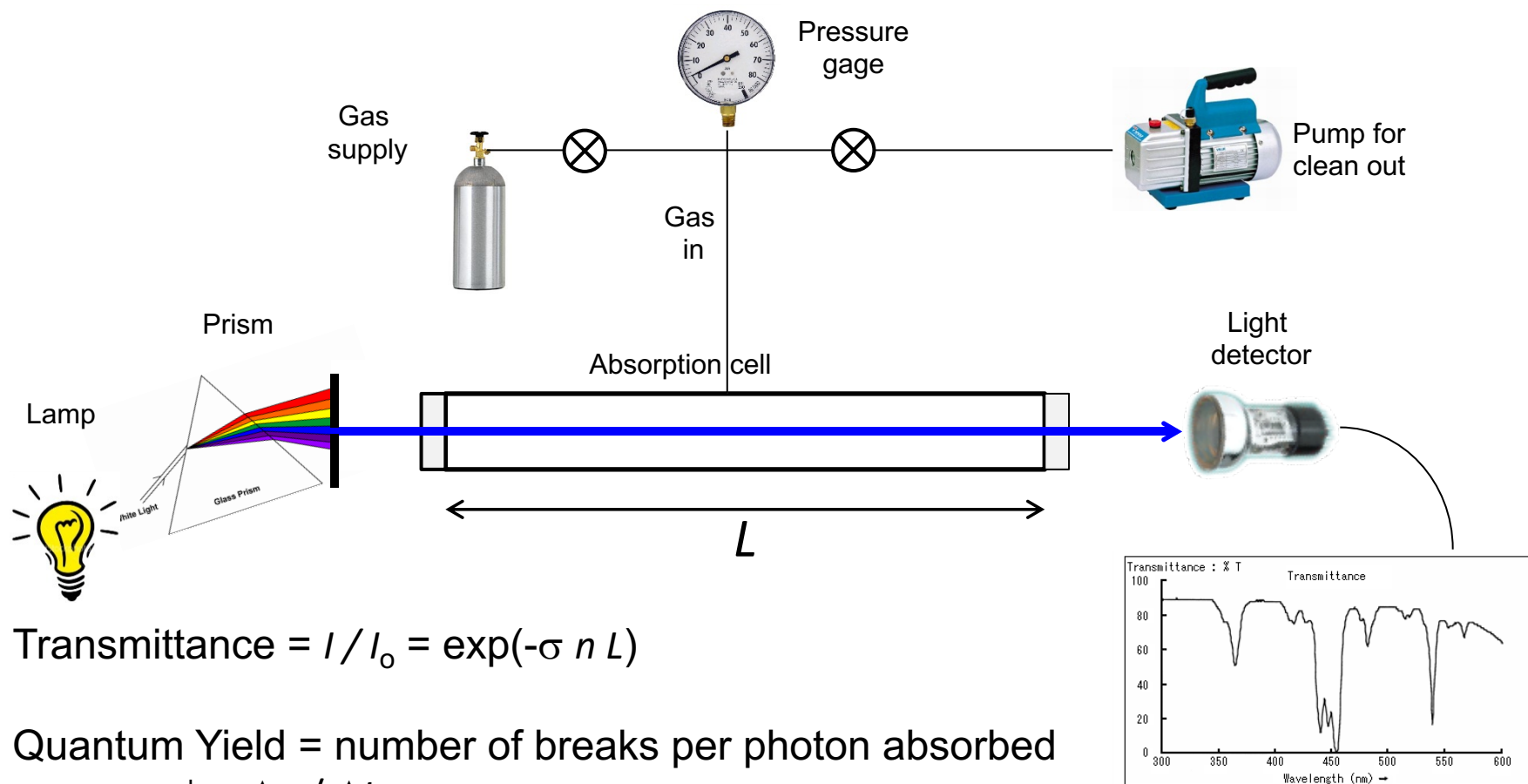


Absorption cross sections of formaldehyde CH_2O at room temperature (results 1990-2003)



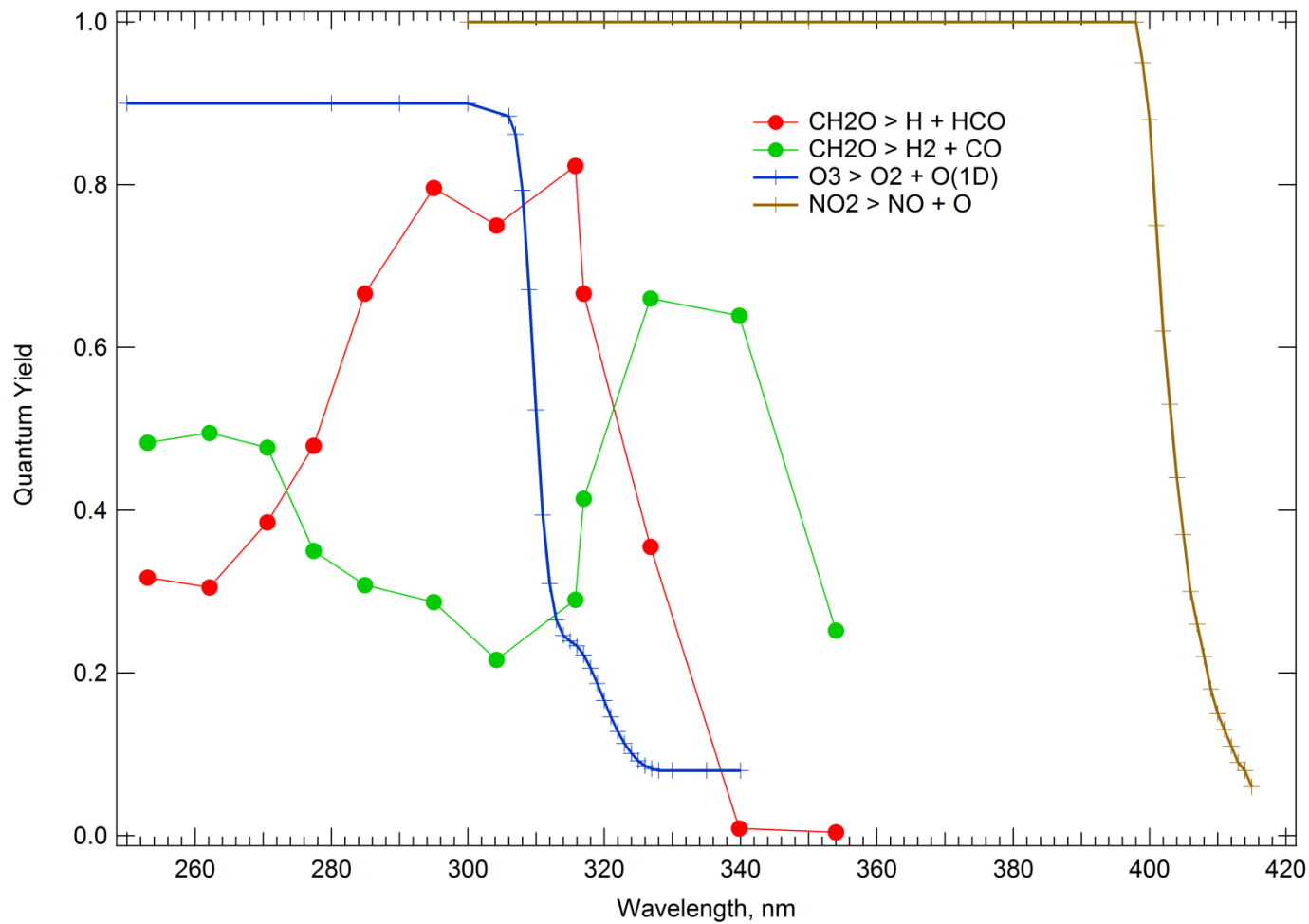
Absorption cross sections of nitrogen dioxide NO_2 at 294 K
Results from the year 1998 and JPL-2006 recommendation

Measurement of Quantum Yields $\phi(\lambda)$



Difficult: must measure absolute change in n (products) and I (photons absorbed)

Photo-dissociation Quantum Yields $\phi(\lambda, T, P)$



Compilations of Cross Sections & Quantum Yields

<http://www.atmosphere.mpg.de/enid/2295>



Max-Planck-Gesellschaft

MPI-Mainz-UV-VIS Spectral Atlas of Gaseous Molecules

A Database of Atmospherically Relevant Species, Including Numerical Data and Graphical Representations

Hannelore Keller-Rudek, Geert K. Moortgat

Max-Planck-Institut für Chemie, Atmospheric Chemistry Division, Mainz, Germany

<http://jpldataeval.jpl.nasa.gov/>

NASA Jet Propulsion Laboratory
California Institute of Technology

[+ View the NASA Portal](#)

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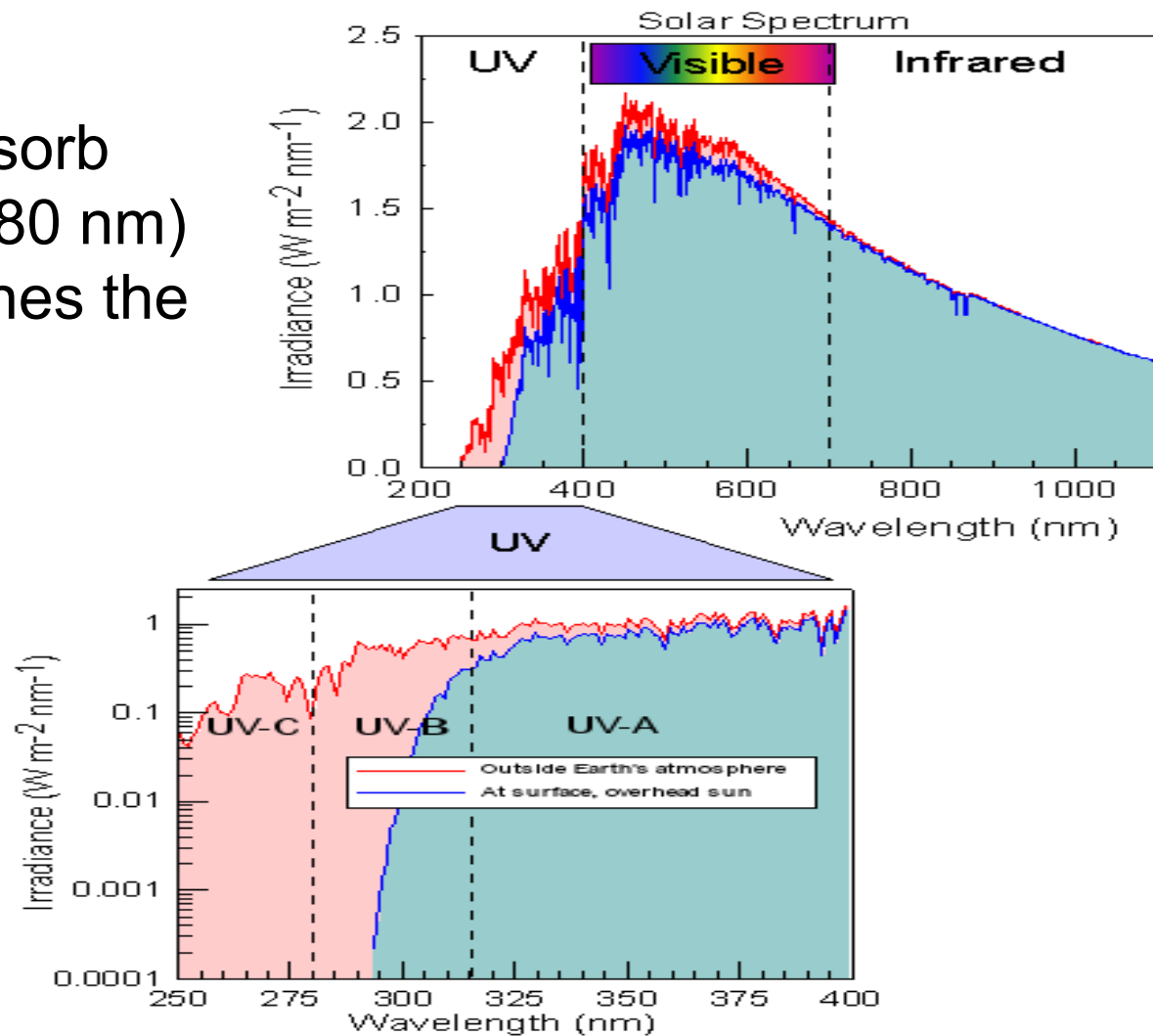
JPL HOME EARTH SOLAR SYSTEM STARS & GALAXIES TECHNOLOGY

NASA/JPL
Data Evaluation

Jet Propulsion Laboratory
California Institute of Technology

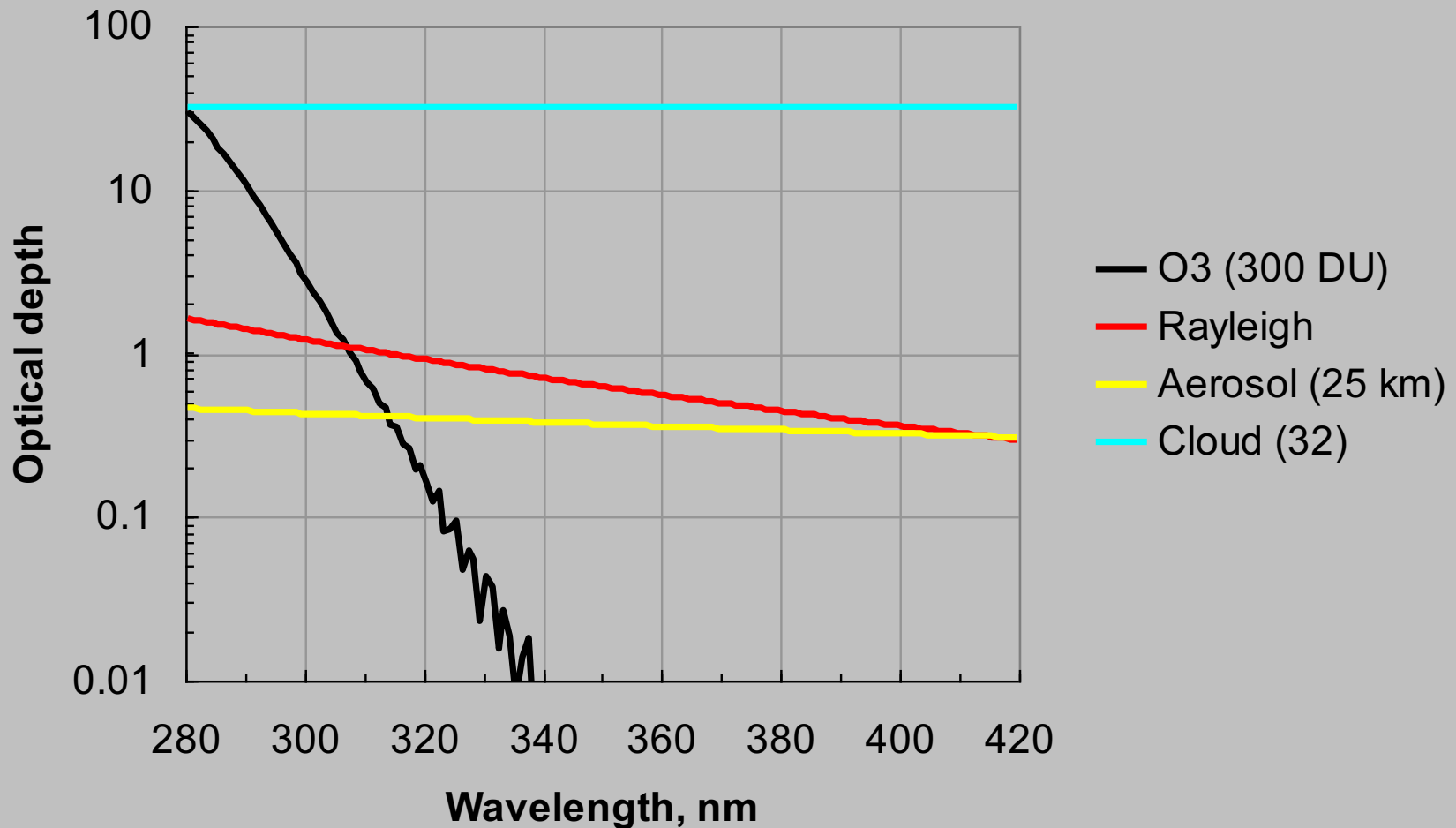
Solar Spectrum

O₂ and O₃ absorb all UV-C ($\lambda < 280$ nm) before it reaches the troposphere



Atmospheric Optical Depths, τ

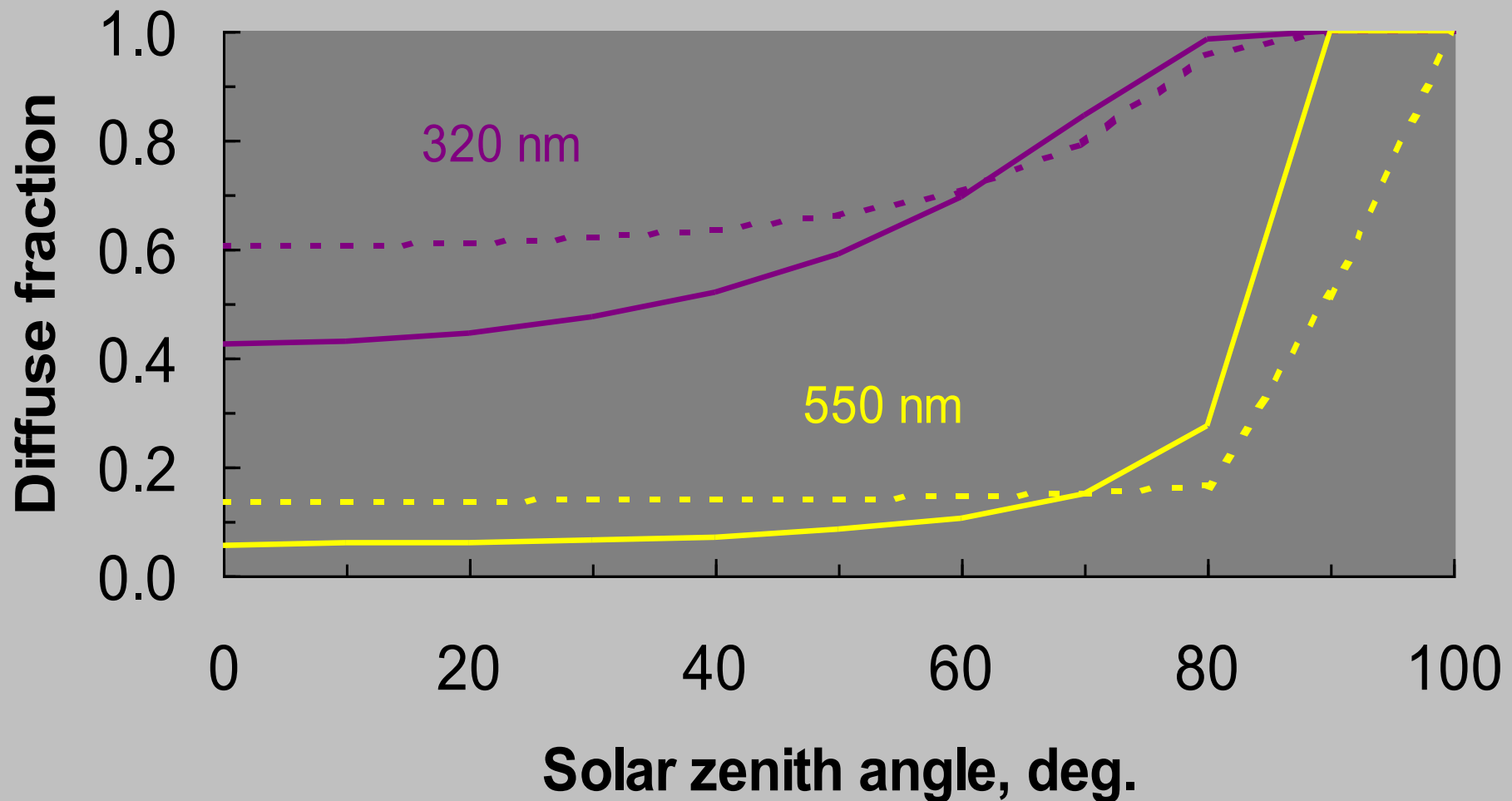
defined by Transmission of a vertical beam = $\exp(-\tau)$



UV: Diffuse Radiation \geq Direct Solar Beam

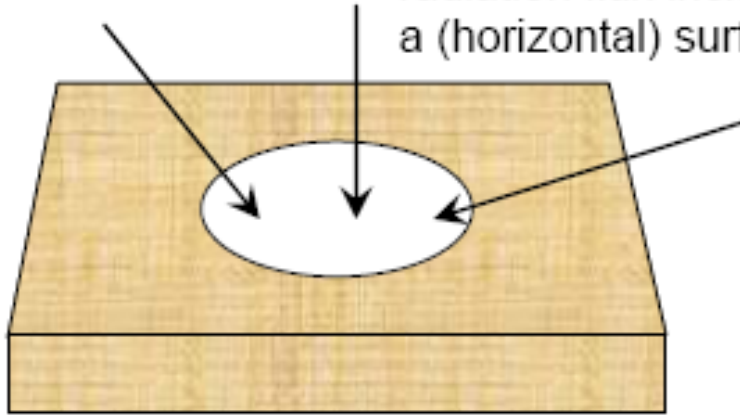
clean skies, sea level

— Irradiance - - - - Actinic flux



INTEGRALS OVER ANGULAR INCIDENCE

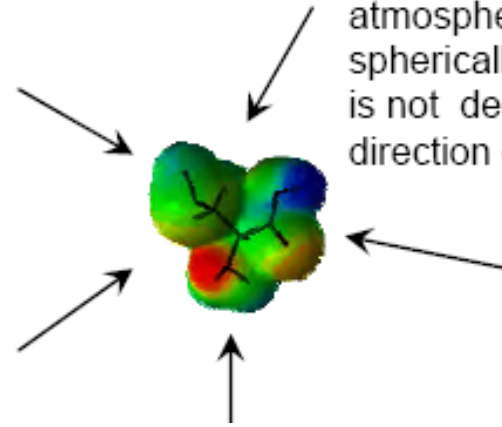
Irradiance: The radiation flux incident on a (horizontal) surface.



$$E = \int_0^{2\pi} \int_0^{\frac{\pi}{2}} I(\theta, \varphi) \cos \theta \sin \theta \, d\theta \, d\varphi$$

Watts m⁻²

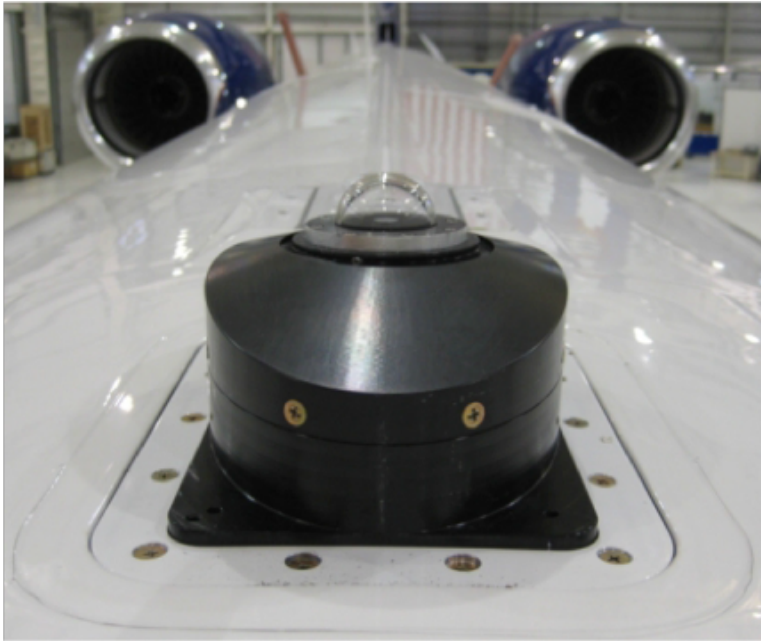
Actinic flux: The photochemically active radiation flux in the earth's atmosphere. This flux is spherically integrated and is not dependent the direction of the radiation.



$$F = \int_0^{\pi} \int_0^{2\pi} I(\theta, \varphi) \sin \theta \, d\varphi \, d\theta$$

Watts m⁻² or quanta s⁻¹ cm⁻²

Actinic Flux Measurements



By placing 2 half spheres on the airplane, you can measure the spectral radiation onto a sphere

*Combine with cross-sections and quantum yields, integrate
→ j-values*



SCATTERING PHASE FUNCTIONS

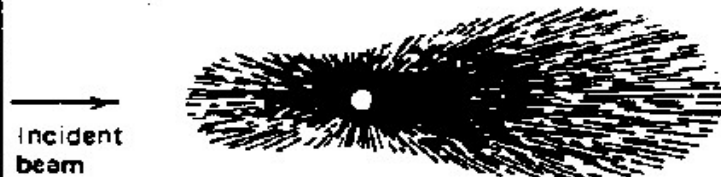
$$P(\theta, \phi; \theta', \phi')$$

Small Particles (a)



Size: smaller than one-tenth the wavelength of light
Description: symmetric

Large Particles (b)



Size: approximately one-fourth the wavelength of light
Description: scattering concentrated in forward direction

Larger Particles (c)



Size: larger than the wavelength of light
Description: extreme concentration of scattering in forward direction;
development of maxima and minima of scattering at
wider angles

The Radiative Transfer Equation

Propagation derivative

*Beer-Lambert
attenuation*

*Scattering from
direct solar beam*

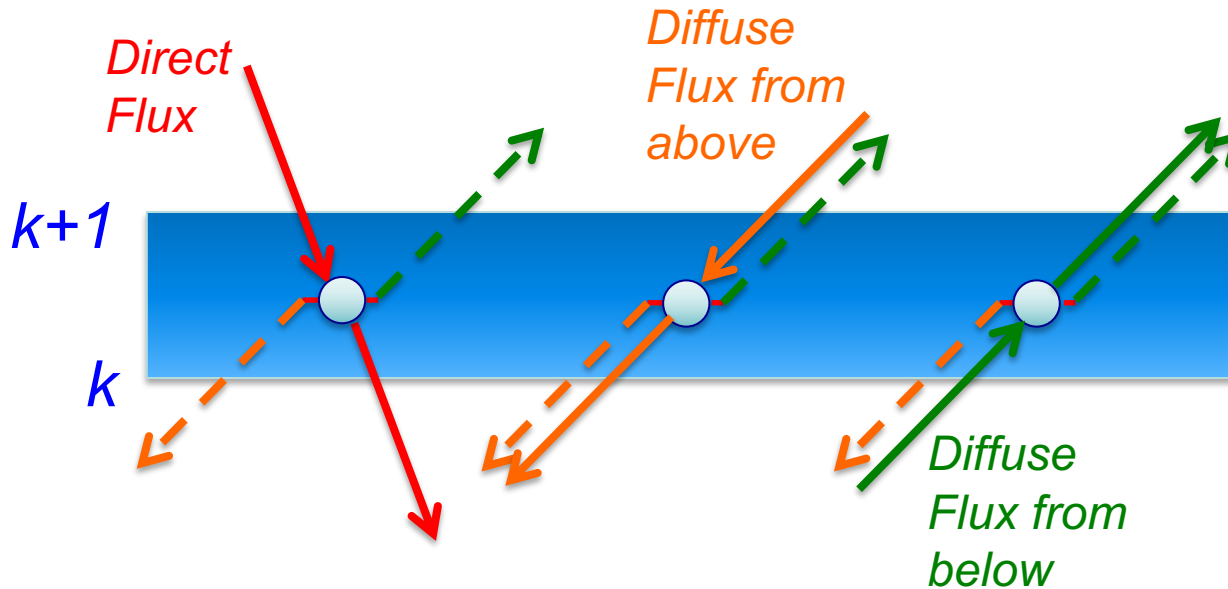
$$\cos \theta \frac{dI(\tau, \theta, \phi)}{d\tau} = -I(\tau, \theta, \phi) + \frac{\omega_o}{4\pi} F_\infty e^{-\tau/\cos \theta_o} P(\theta, \phi; \theta_o, \phi_o) + \frac{\omega_o}{4\pi} \int_0^{2\pi} \int_{-1}^{+1} I(\tau, \theta', \phi') P(\theta, \phi; \theta', \phi') d\cos \theta' d\phi'$$

*Scattering from diffuse light
(multiple scattering)*

NUMERICAL SOLUTIONS TO RADIATIVE TRANSFER EQUATION

- **Discrete ordinates**
n-streams ($n = \text{even}$), angular distribution
exact as $n \rightarrow \infty$ but speed $\propto 1/n^2$
- **Two-stream family**
delta-Eddington, many others
very fast but not exact
- **Monte Carlo**
slow, but ideal for 3D problems
- **Others**
matrix operator, Feautrier, adding-doubling,
successive orders, etc.

Two-stream methods



Multiple atmospheric layers, each assumed to be homogeneous
Must specify three optical properties:

Optical depth, $\Delta\tau$

Single scattering albedo, $\omega_o = \text{scatt.}/(\text{scatt.}+\text{abs.})$

Asymmetry factor, g : forward fraction $\sim (1+g)/2$

For each layer, must specify $\Delta\tau$, ω_o , g :

1. Vertical optical depth, $\Delta\tau(\lambda, z) = \sigma(\lambda, z) n(z) \Delta z$

for molecules: $\Delta\tau(\lambda, z) \sim 0 - 30$

Rayleigh scatt. $\sim 0.1 - 1.0 \sim \lambda^{-4}$

O₃ absorption $\sim 0 - 30$

for aerosols: 0.01 - 5.0

Mie scatt. $\Delta\tau(\lambda, z) \sim \lambda^{-\alpha}$

(α = Angstrom exponent)

for clouds: 1-1000

$\alpha \sim 0$

cirrus $\sim 1-5$

cumulonimbus $\sim > 100$

For each layer, must specify $\Delta\tau$, ω_o , g :

2. Single scattering albedo, $\omega_o(\lambda, z) = \text{scatt.}/(\text{scatt.}+\text{abs.})$

range 0 - 1

limits: pure scattering = 1.0
pure absorption = 0.0

for molecules, strongly λ -dependent, depending on
absorber amount, esp. O_3

for aerosols:

sulfate ~ 0.99
soot, organics ~ 0.8 or less,
not well known but probably higher
at shorter λ , esp. in UV

for clouds: typically 0.9999 or larger (vis and UV)

For each layer, must specify $\Delta\tau$, ω_o , g :

3. Asymmetry factor, $g(\lambda, z)$ = first moment of phase function

range -1 to + 1

pure back-scattering = -1

isotropic or Rayleigh = 0

pure forward scattering = +1

$$g = \frac{1}{2} \int_{-1}^{+1} P(\Theta) \cos \Theta d(\cos \Theta)$$

strongly dependent on particle size

for aerosols:, typically 0.5-0.7

for clouds, typically 0.7-0.9

*Mie theory for spherical particles: can compute $\Delta\tau$, ω_o , g
from knowledge of λ , particle radius and complex index of refraction*

Mie Scattering Theory

For spherical particles, given:

Complex index of refraction: $n = m + ik$
(composition-dependent)

Size parameter: $\alpha = 2\pi r / \lambda$

Can compute:

Extinction efficiency $Q_e(\alpha, n) \times \pi r^2$

Scattering efficiency $Q_s(\alpha, n) \times \pi r^2$

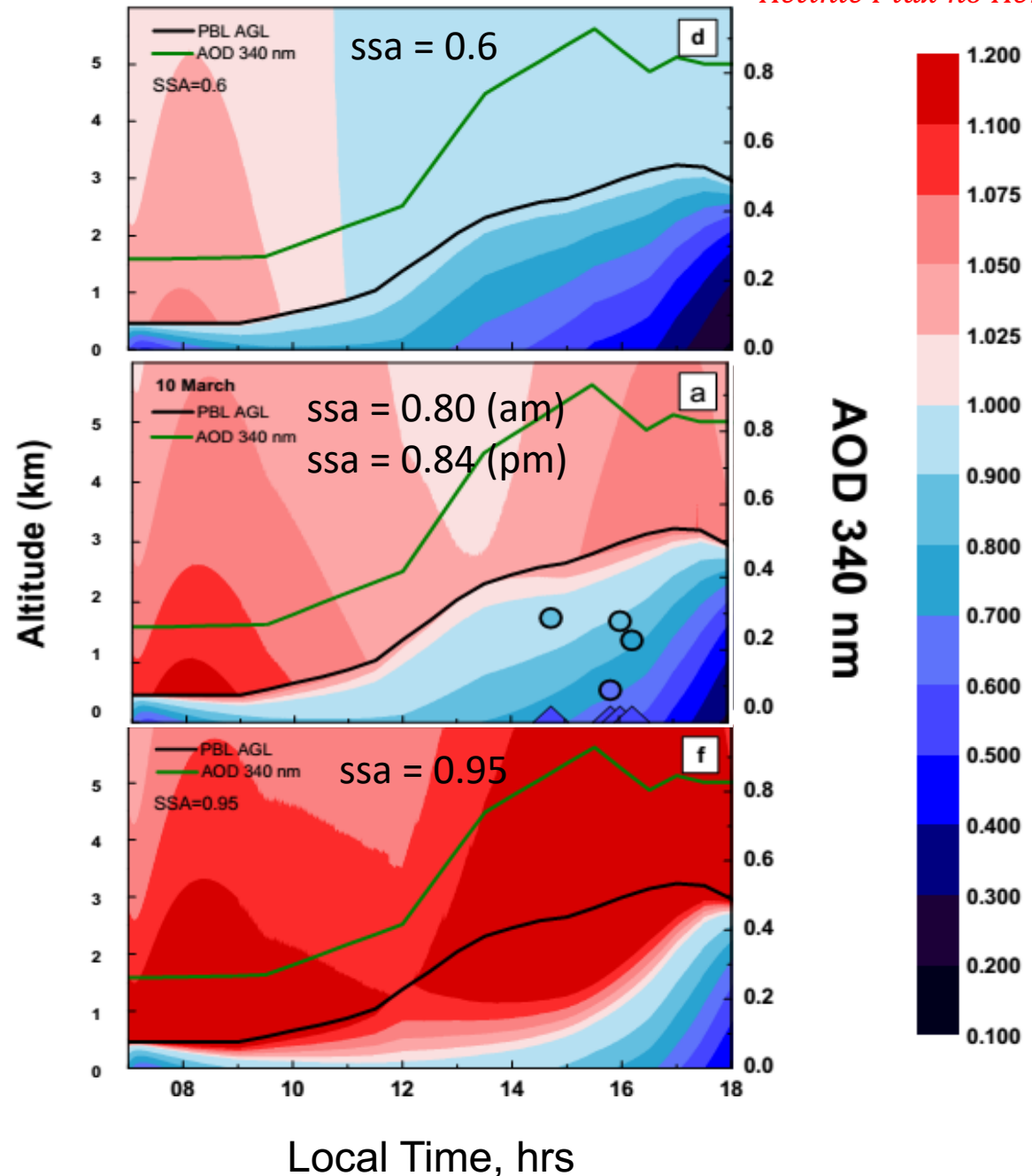
Phase function $P(\Theta, \alpha, n)$
or asymmetry factor $g(\alpha, n)$

Actinic Flux is Very Sensitive to Single Scattering Albedo

Mexico City suburbs (T1)
March 2006

Central panel:
Model with observed
ssa, and obs.

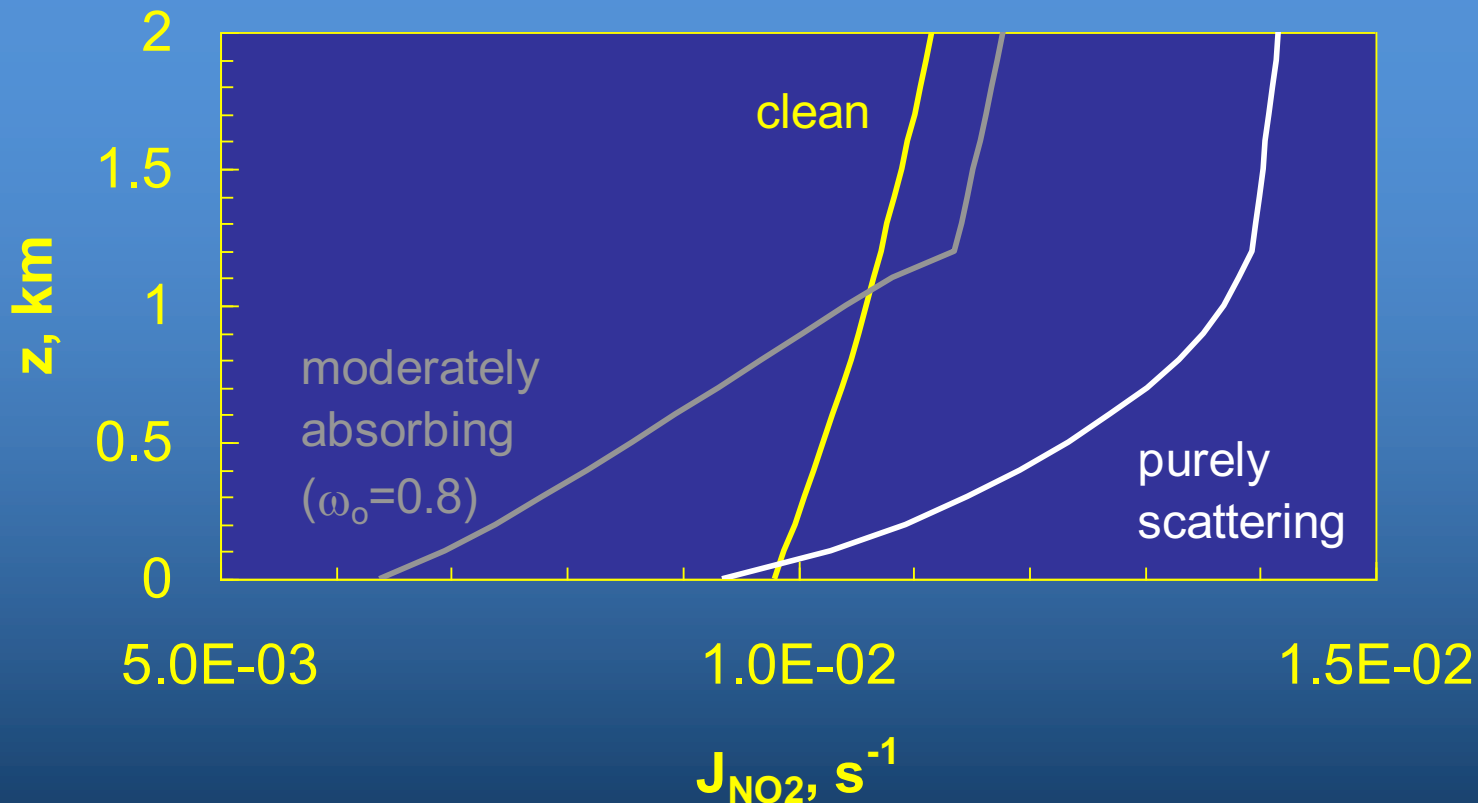
Upper and lower
panels: Sensitivity to
ssa



Effects on the Radiation

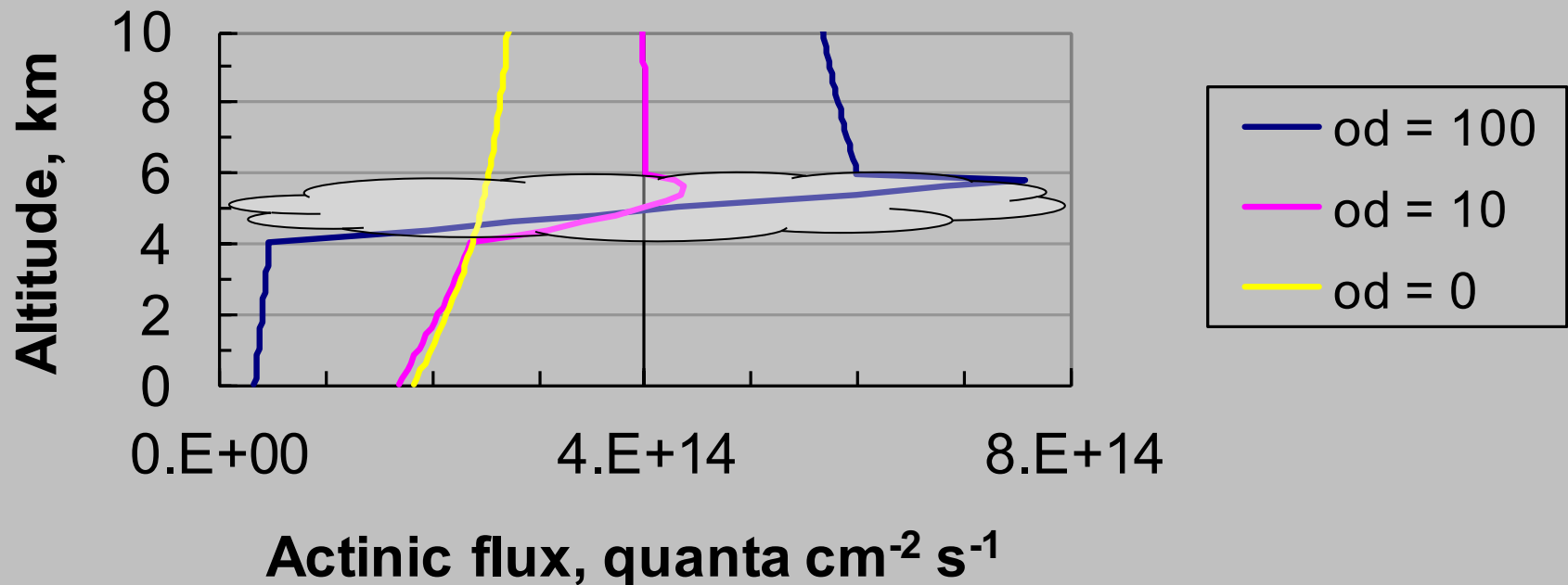
- Aerosols affect radiation – either scatter or absorb
- Clouds affect radiation – scattering

*NO₂ Photolysis Frequency
19N, April, noon, AOD = 1 at 380 nm*

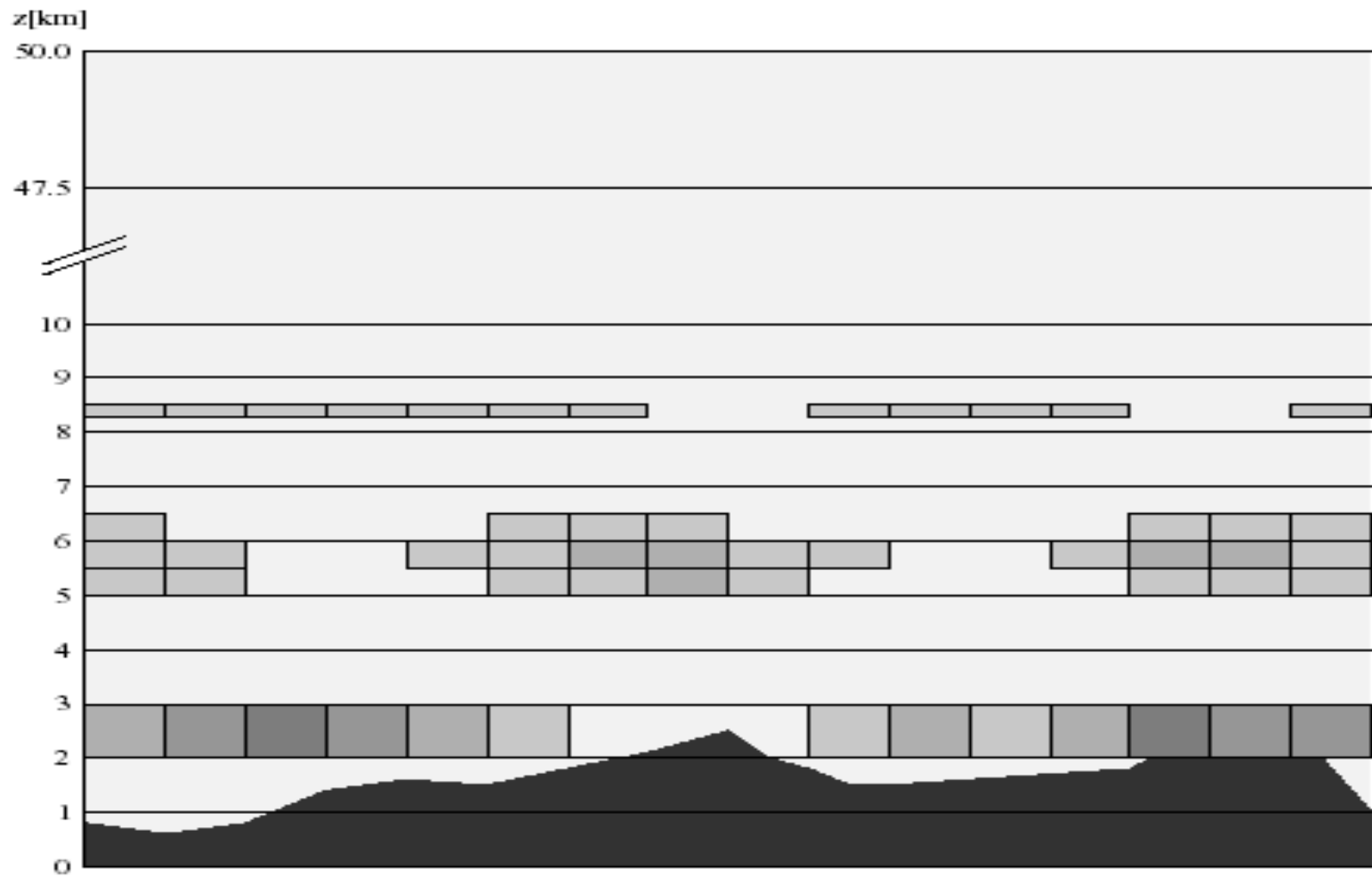


EFFECT OF UNIFORM CLOUDS ON ACTINIC FLUX

**340 nm, sza = 0 deg.,
cloud between 4 and 6 km**



Broken Clouds



Photolysis in WRF-Chem

- Several radiative transfer options:
 - phot_opt = 1 : Madronich TUV (140 λ s, delta-Eddington)
 - phot_opt = 2 : Fast-J (17 λ s, 8-str Feautrier)
 - phot_opt = 3 : F-TUV (17 λ s, correction factor, delta-Eddington)
 - ⇒ phot_opt = 4: updated TUV (140 λ s, delta-Eddington)
- New option in WRF-Chem v3.9:
 - ⇒ only works with MOZART_MOSAIC_4BIN_KPP, MOZART_MOSAIC_4BIN_AQ_KPP, and MOZCART_KPP chemical options
- Limitations & advantages
 - Cross section and quantum yield data are hard-coded and not up to date in older schemes;
 - ⇒ phot_opt = 4 has updated database to the latest TUV model (V5.3, Oct. 2016)
 - Difficult to add new reactions (typically available ~ 20)
 - ⇒ 109 reactions relevant for tropospheric & stratospheric chemistry

List of available photolysis reactions in the updated TUV

1	$O_2 \rightarrow O + O$	(J_o2)	31	$2-C_4H_9ONO_2 \rightarrow 2-C_4H_9O + NO_2$	
2	$O_3 \rightarrow O_2 + O(1D)$	(J_o1d)	32	$CH_3CHONO_2CH_3 \rightarrow CH_3CHOCH_3 + NO_2$	
3	$O_3 \rightarrow O_2 + O(3P)$	(J_o3p)	33	$CH_2(OH)CH_2(ONO_2) \rightarrow CH_2(OH)CH_2(O.) + NO_2$	
4	$HO_2 \rightarrow OH + O$		34	$CH_3COCH_2(ONO_2) \rightarrow CH_3COCH_2(O.) + NO_2$	
5	$H_2O_2 \rightarrow 2 OH$	(J_h2o2)	35	$C(CH_3)_3(ONO_2) \rightarrow C(CH_3)_3(O.) + NO_2$	
6	$NO_2 \rightarrow NO + O(3P)$	(J_no2)	36	$C(CH_3)_3(ONO) \rightarrow C(CH_3)_3(O) + NO$	
7	$NO_3 \rightarrow NO + O_2$		37	$CH_3CO(OONO_2) \rightarrow CH_3CO(OO) + NO_2$	(J_pan_a)
8	$NO_3 \rightarrow NO_2 + O(3P)$		38	$CH_3CO(OONO_2) \rightarrow CH_3CO(O) + NO_3$	(J_pan_b)
9	$N_2O \rightarrow N_2 + O(1D)$	(J_n2o)	39	$CH_3CH_2CO(OONO_2) \rightarrow CH_3CH_2CO(OO) + NO_2$	
10	$N_2O_5 \rightarrow NO_3 + NO + O(3P)$		40	$CH_3CH_2CO(OONO_2) \rightarrow CH_3CH_2CO(O) + NO_3$	
11	$N_2O_5 \rightarrow NO_3 + NO_2$	(J_n2o5b)	41	$CH_2=CHCHO \rightarrow$ Products	
12	$HNO_2 \rightarrow OH + NO$		42	$CH_2=C(CH_3)CHO \rightarrow$ Products	(J_macr)
13	$HNO_3 \rightarrow OH + NO_2$	(J_hno3)	43	$CH_3COCH=CH_2 \rightarrow$ Products	(J_mvk)
14	$HNO_4 \rightarrow HO_2 + NO_2$	(J_hno4)	44	$HOCH_2CHO \rightarrow CH_2OH + HCO$	(J_glyald_a)
15	$NO_3-(aq) \rightarrow NO_2(aq) + O-$		45	$HOCH_2CHO \rightarrow CH_3OH + CO$	(J_glyald_b)
16	$NO_3-(aq) \rightarrow NO_2-(aq) + O(3P)$		46	$HOCH_2CHO \rightarrow CH_2CHO + OH$	(J_glyald_c)
17	$CH_2O \rightarrow H + HCO$	(J_ch2or)	47	$CH_3COCH_3 \rightarrow CH_3CO + CH_3$	(J_ch3coch3)
18	$CH_2O \rightarrow H_2 + CO$	(J_ch2om)	48	$CH_3COCH_2CH_3 \rightarrow CH_3CO + CH_2CH_3$	(J_mek)
19	$CH_3CHO \rightarrow CH_3 + HCO$	(J_ch3cho_a)	49	$CH_2(OH)COCH_3 \rightarrow CH_3CO + CH_2(OH)$	(J_hyac_a)
20	$CH_3CHO \rightarrow CH_4 + CO$	(J_ch3cho_b)	50	$CH_2(OH)COCH_3 \rightarrow CH_2(OH)CO + CH_3$	(J_hyac_b)
21	$CH_3CHO \rightarrow CH_3CO + H$	(J_ch3cho_c)	51	$CHOCHO \rightarrow HCO + HCO$	(J_gly_a)
22	$C_2H_5CHO \rightarrow C_2H_5 + HCO$		52	$CHOCHO \rightarrow H_2 + 2CO$	(J_gly_b)
23	$CH_3OOH \rightarrow CH_3O + OH$		53	$CHOCHO \rightarrow CH_2O + CO$	(J_gly_c)
24	$HOCH_2OOH \rightarrow HOCH_2O. + OH$	(J_pooh)	54	$CH_3COCHO \rightarrow CH_3CO + HCO$	(J_mgly)
25	$CH_3ONO_2 \rightarrow CH_3O + NO_2$		55	$CH_3COCOCH_3 \rightarrow$ Products	
26	$CH_3(OONO_2) \rightarrow CH_3(OO) + NO_2$		56	$CH_3COOH \rightarrow CH_3 + COOH$	
27	$CH_3CH_2ONO_2 \rightarrow CH_3CH_2O + NO_2$		57	$CH_3CO(OOH) \rightarrow$ Products	
28	$C_2H_5ONO_2 \rightarrow C_2H_5O + NO_2$		58	$CH_3COCO(OH) \rightarrow$ Products	
29	$n-C_3H_7ONO_2 \rightarrow C_3H_7O + NO_2$		59	$(CH_3)_2NNO \rightarrow$ Products	<i>*in mozart_mosaic_4bin</i>
30	$1-C_4H_9ONO_2 \rightarrow 1-C_4H_9O + NO_2$		60	$CF_2O \rightarrow$ Products	

List of available photolysis reactions in the updated TUV

61	Cl ₂ -> Cl + Cl	91	CF ₃ CF ₂ CHCl ₂ (HCFC-225ca) -> Products
62	ClO -> Cl + O(1D)	92	CF ₂ ClCF ₂ CHFCl (HCFC-225cb) -> Products
63	ClO -> Cl + O(3P)	93	Br ₂ -> Br + Br
64	ClOO -> Products	94	BrO -> Br + O
65	OCIO -> Products	95	HOBr -> OH + Br
66	ClOOCl -> Cl + ClOO	96	BrNO -> Br + NO
67	HCl -> H + Cl	97	BrONO -> Br + NO ₂
68	HOCl -> HO + Cl	98	BrONO -> BrO + NO
69	NOCl -> NO + Cl	99	BrNO ₂ -> Br + NO ₂
70	ClNO ₂ -> Cl + NO ₂	100	BrONO ₂ -> BrO + NO ₂
71	ClONO -> Cl + NO ₂	101	BrONO ₂ -> Br + NO ₃
72	ClONO ₂ -> Cl + NO ₃	102	BrCl -> Br + Cl
73	ClONO ₂ -> ClO + NO ₂	103	CH ₃ Br -> Products
74	CCl ₄ -> Products	104	CHBr ₃ -> Products
75	CH ₃ OCl -> CH ₃ O + Cl	105	CF ₂ Br ₂ (Halon-1202) -> Products
76	CHCl ₃ -> Products	106	CF ₂ BrCl (Halon-1211) -> Products
77	CH ₃ Cl -> Products	107	CF ₃ Br (Halon-1301) -> Products
78	CH ₃ CCl ₃ -> Products	108	CF ₂ BrCF ₂ Br (Halon-2402) -> Products
79	CCl ₂ O -> Products	109	perfluoro 1-iodopropane -> products
80	CClFO -> Products		
81	CCl ₃ F (CFC-11) -> Products		
82	CCl ₂ F ₂ (CFC-12) -> Products		
83	CF ₂ ClCFCl ₂ (CFC-113) -> Products		
84	CF ₂ ClCF ₂ Cl (CFC-114) -> Products		
85	CF ₃ CF ₂ Cl (CFC-115) -> Products		
86	CHClF ₂ (HCFC-22) -> Products		
87	CF ₃ CHCl ₂ (HCFC-123) -> Products		
88	CF ₃ CHFCl (HCFC-124) -> Products		
89	CH ₃ CFCl ₂ (HCFC-141b) -> Products		
90	CH ₃ CF ₂ Cl (HCFC-142b) -> Products		

*Additional file in KPP/mechanisms/\$mechanism/
\$mechanism.tuv.jmap*

Provides mapping of j_wrfchem with available j_tuv

Photolysis in WRF-Chem

- Ozone column density above the model top:
 - 1) TUV: specified value above the model top (specified_du=325)
 - 2) fast-J: specified value at the model top for the whole domain
 - 3) f-TUV: WACCM model climatology at the top (input file exo_coldens.nc)
 - 4) New TUV: uses ozone climatology distributed from model top to 50km, and then several options available above 50km
- Cloud optical properties:
 - Recalculated in each photolysis scheme, different from physics (e.g. RRTMG)
 - typically, COD calculated from LWP/IWP and effective drop radius (Slingo 1989, with fixed SSA = 0.9999 and $f_{\text{assym}} = 0.85$)
 - Various treatments of Sub-grid cloud overlap
 - Scaled by cloud fraction (fast-J)
 - Max random overlap for f-TUV (expensive)
 - Simplified ($\text{COD}_{\text{subgrid}} = \text{COD} * \text{FCLD}^{3/2}$, equivalent to max random overlap)
- Aerosols:
accounted for through the namelist option **aer_ra_feedback = .true.**

Settings for phot_opt = 4 (default in red)

Download the data file TUV.phot.tar from the ACOM website
(add data directories DATAE1 and DATAJ1, and wrf_tuv_xsqr.nc file)

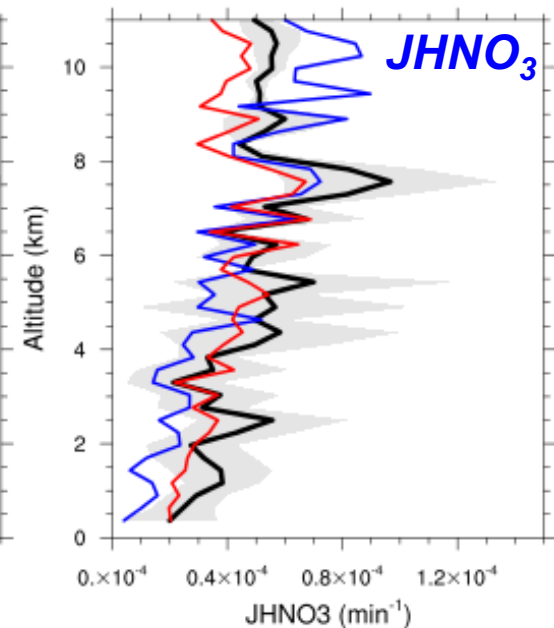
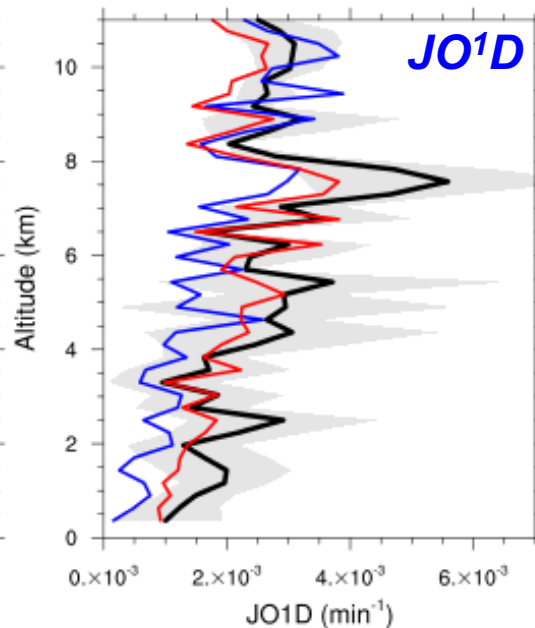
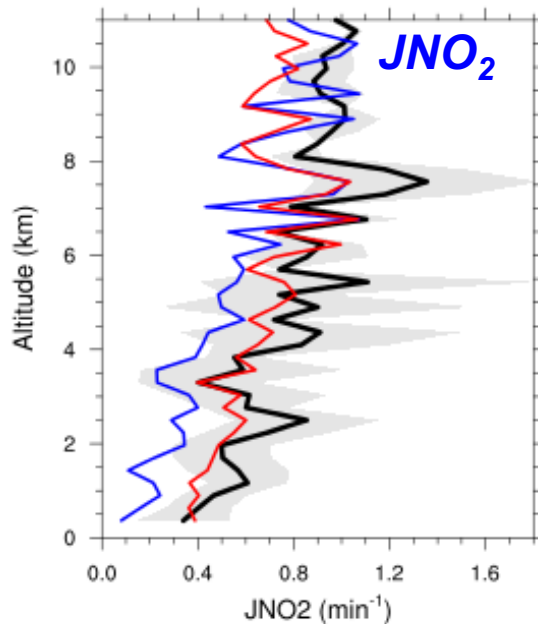
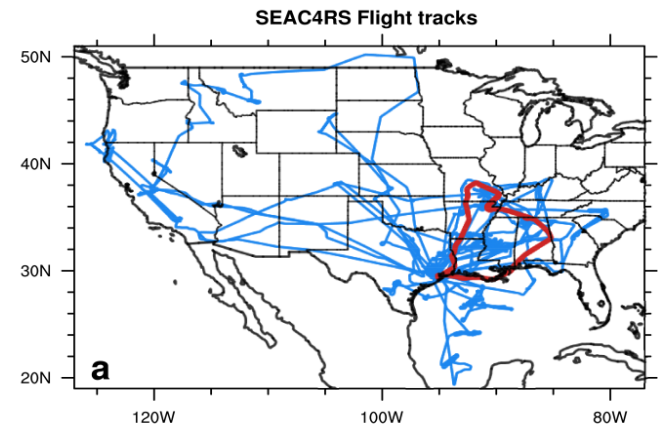
- phot_opt = 4, 4
- is_full_tuv = .false. : use wrf_tuv_xsqr.nc table interpolation
- is_full_tuv = .true. : use hard-coded data and formulas (updated)
- du_at_grnd = 300 : default total o3 column density
- has_o3_exo_coldens = .false. : o3 column density above 50 km = 0.
- has_o3_exo_coldens = .true. : o3 column density above 50 km from WACCM climatology
- scale_o3_to_grnd_exo_coldens = .true. : total o3 column at ground scaled to climatology
- scale_o3_to_du_at_grnd = .true. : scaled to the du_at_grnd value at the ground
- pht_cldfrc_opt = 1 : grid cell cloud fraction is either 0 or 1
- pht_cldfrc_opt = 2 : grid cell cloud fraction varies between 0 and 1
- cld_od_opt = 1 : cloud optical depth is scaled by cloud fraction
- cld_od_opt = 2 : cloud optical depth is scaled by (cloud fraction)**1.5

Comparison with the 2013 SEAC⁴RS flights

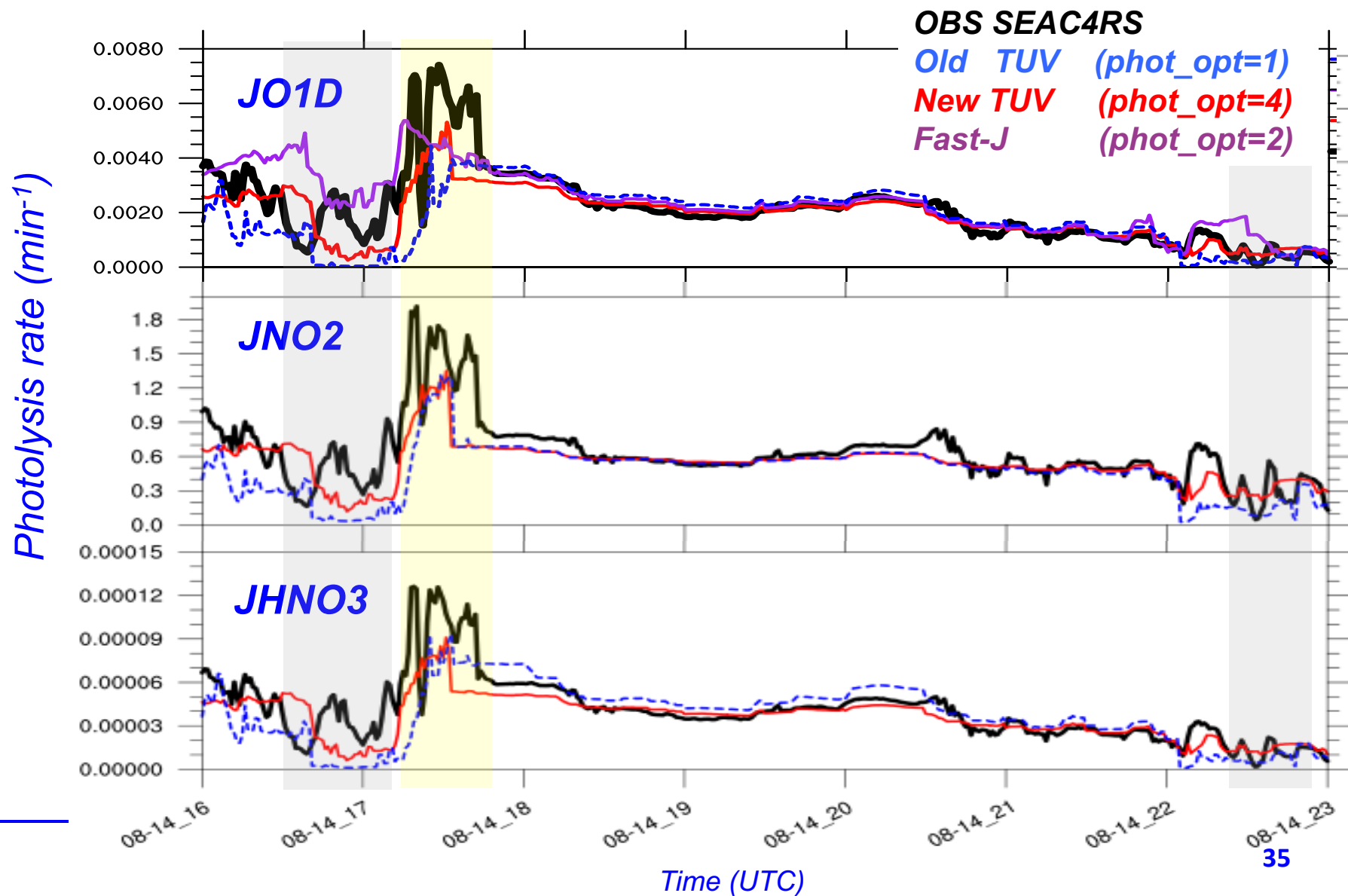
OBS SEAC4RS

Old TUV (*phot_opt=1*)

New TUV (*phot_opt=4*)



Comparison with SEAC⁴RS (14 Aug. 2013)



QUICK TUV CALCULATOR

[ACD](#) > [Models](#) > [TUV](#) > [Interactive TUV](#)

This web page runs the 4.1 version of the TUV model. You can run the model for a specified latitude, longitude and time (input option 1), or for a given solar zenith angle (input option 2). In either case, you must also specify the additional parameters in the second column. Also, you may select to print out the photolysis rates and/or the solar actinic flux spectrum at a given altitude above the surface (output option 1), or the erythemal UV and/or solar irradiance at that altitude (output option 2). For any problem, or to send comments, email [TUV administrators](#).

<p><input checked="" type="radio"/> INPUT OPTION 1</p> <p>LATITUDE (deg): <input type="text" value="0"/></p> <p>LONGITUDE (deg): <input type="text" value="0"/></p> <p>TIME (hh:mm:ss, GMT): <input type="text" value="12:00:00"/></p> <p><input type="radio"/> INPUT OPTION 2</p> <p>SOLAR ZENITH ANGLE <input type="text" value="0"/> (deg):</p>	<p>OTHER INPUT PARAMETERS</p> <p>DATE (YYMMDD): <input type="text" value="000630"/></p> <p>OVERHEAD OZONE COLUMN <input type="text" value="300"/> (du):</p> <p>SURFACE ALBEDO (0-1): <input type="text" value="0.1"/></p> <p>GROUND ELEVATION (km asl): <input type="text" value="0"/></p> <p>MEASUREM. ALTITUDE (km <input type="text" value="0"/> asl):</p>	<p><input checked="" type="radio"/> OUTPUT OPTION 1 (for Atmospheric Science)</p> <p><input checked="" type="checkbox"/> MOLECULAR PHOTOLYSIS FREQUENCIES (s-1)</p> <p><input type="checkbox"/> ACTINIC FLUX, SPECTRAL (quanta s-1 cm-2 nm-1)</p> <p><input type="radio"/> OUTPUT OPTION 2 (for Biology)</p> <p><input checked="" type="checkbox"/> IRRADIANCE, WEIGHTED (W m-2)</p> <p><input type="checkbox"/> IRRADIANCE, SPECTRAL (W m-2 nm-1)</p>
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RADIATION TRANSFER MODEL

- ☒ Pseudo-spherical 2 streams (faster, less accurate)
- ☐ Pseudo-spherical discrete ordinate 4 streams (slower, more accurate)

[GO!](#)